Vocabulary #8 STAAR Matter and Energy Lesson 2: Mixtures and Solutions

Boiling point	The temperature at which a substance changes from a liquid to a gas.	Bubbles contain water in gas state. Vapor pressure is equal or greater than atmospheric pressure. C. Ophardt, c. 2003	The boiling point of water is 100 degrees Celsius.
Melting point	The temperature at which as substance changes from a solid to a liquid.	Melting Point 32° Fahrenheit 0° Celsius	The melting point of water is 0 degrees Celsius.
Physical property	A characteristic of matter that can be observed or measured and can be changed without changing the matter itself.		One physical property of water is that it occurs naturally on earth in all three states. We can watch it change from a solid to a liquid to a gas.
Density	Refers to whether an object will sink or float.		I can tell the cork is less dense than the water because it floats on the water's surface.

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Mixture	A combination of two or more substances that do not join together to form a new substance.		Marbles and wooden blocks together in a container would be a mixture.
Solubility	The ability of a substance to dissolve in a liquid.	BALL	The solubility of the powder was tested as we watched it dissolve in the liquid.
Solution	A mixture in which the substances are spread out evenly between one another and cannot be told apart.		Sugar and water form a solution when mixed together.

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Freezing point	The temperature at which a substance changes from a liquid to a solid.	\$30°C	Water will reach its freezing point at 0 degrees Celsius.
Physical change	A change in the appearance of matter without actually changing the matter itself.	Physical Change	Cutting a piece of cardboard would be a physical change.
Temperature	The measurement of how hot or cold a substance is.		The students were required to measure the temperature of the liquid every 3 minutes for 12 minutes.